## Cambridge International AS \& A Level



CENTRE NUMBER


CANDIDATE NUMBER

## BIOLOGY

You must answer on the question paper.
You will need: The materials and apparatus listed in the confidential instructions

## INSTRUCTIONS

- Answer all questions.
- Use a black or dark blue pen. You may use an HB pencil for any diagrams or graphs.
- Write your name, centre number and candidate number in the boxes at the top of the page.
- Write your answer to each question in the space provided.
- Do not use an erasable pen or correction fluid.
- Do not write on any bar codes.
- You may use a calculator.
- You should show all your working and use appropriate units.


## INFORMATION

- The total mark for this paper is 40 .
- The number of marks for each question or part question is shown in brackets [ ].

| For Examiner's Use |  |
| :---: | :---: |
| 1 |  |
| 2 |  |
| Total |  |

This document has 16 pages. Any blank pages are indicated.

Before you proceed, read carefully through the whole of Question 1 and Question 2.
Plan the use of the two hours to make sure that you finish the whole of Question 1 and Question 2.

1 Plant cells contain the enzyme catalase which catalyses the breakdown of hydrogen peroxide, releasing oxygen.

When a mixture of hydrogen peroxide and plant extract is put into a syringe, bubbles of oxygen are released.

You are going to investigate the effect of substrate concentration on the activity of catalase.
You are provided with the materials shown in Table 1.1.
Table 1.1

| labelled | contents | hazard | volume/cm $\mathbf{c o}^{\mathbf{3}}$ |
| :---: | :--- | :---: | :---: |
| H | $3.0 \%$ hydrogen peroxide solution | irritant | 30 |
| P | plant extract solution | none | 40 |
| U | unknown concentration of hydrogen peroxide solution | irritant | 10 |
| W | distilled water | none | 100 |

If any of the solutions come into contact with your skin, wash off immediately with cold water.
It is recommended that you wear suitable eye protection and wear gloves to protect your hands when using hydrogen peroxide, $\mathbf{H}$ and $\mathbf{U}$.
(a) You will need to carry out a serial dilution of the $3.0 \%$ hydrogen peroxide solution, $\mathbf{H}$, to reduce the concentration by half between each successive dilution.

You will need to prepare four concentrations of hydrogen peroxide solution in addition to the $3.0 \%$ hydrogen peroxide solution, $\mathbf{H}$.

After the serial dilution is completed, you will need to have $10 \mathrm{~cm}^{3}$ of each concentration available to use.
(i) Complete Fig. 1.1 to show how you will prepare your serial dilution.

Fig. 1.1 shows the first two beakers you will use to make your serial dilution. You will need to draw three additional beakers.

For each beaker add labelled arrows to show:

- the volume of hydrogen peroxide solution transferred
- the volume of distilled water, W, added.

Under each beaker, state the concentration of hydrogen peroxide solution.

$\qquad$
$\qquad$
$\qquad$
$\qquad$

Fig. 1.1

Carry out step 1 to step 24.

1. Prepare the concentrations of hydrogen peroxide solution, as decided in (a)(i), in the beakers provided.
2. Put $\mathbf{W}$ into the test-tube so that it is approximately half-full.
3. Use the glass rod to stir the plant extract solution, $\mathbf{P}$.
4. Put the nozzle of a clean syringe into the beaker containing $\mathbf{P}$.
5. Pull the plunger out to the $1 \mathrm{~cm}^{3}$ mark so that $\mathbf{P}$ enters the syringe, as shown in Fig. 1.2.


Fig. 1.2
6. Remove the syringe from the beaker containing $\mathbf{P}$ and wipe the nozzle with a paper towel.
7. Put the nozzle of the same syringe into the beaker containing $3.0 \%$ hydrogen peroxide solution, $\mathbf{H}$.
8. Pull the plunger out to the $2 \mathrm{~cm}^{3}$ mark so that $1 \mathrm{~cm}^{3}$ of the $3.0 \%$ hydrogen peroxide solution enters the syringe, as shown in Fig. 1.3.


Fig. 1.3
9. Carefully wipe the nozzle with a paper towel to remove excess hydrogen peroxide solution.
10. Put the tubing onto the syringe nozzle.
11. Put the syringe into a beaker as shown in Fig. 1.4.


Fig. 1.4
12. Put the end of the delivery tube into the test-tube as shown in Fig. 1.4.
13. Start timing when the first bubble is observed in the water in the test-tube.
14. Count the number of bubbles produced in 120 seconds. Record the results in (a)(ii).
15. Hold the syringe with the nozzle up and remove the tubing from the syringe. Put the tubing onto a paper towel.
16. Hold the syringe with the nozzle pointing downwards over the container labelled 'For waste' and empty the syringe.
17. Fill the syringe with water from the container labelled 'For washing'.
18. Put the tubing back onto the syringe nozzle.
19. Empty the syringe through the tubing into the container labelled 'For waste'.
20. Repeat step 15 to step 19 twice more to wash the syringe and the tubing.
21. Remove the tubing from the syringe. Put the tubing onto a paper towel.
22. Repeat step 3 to step 21 with each of the other concentrations of hydrogen peroxide solution prepared in step 1.
(ii) Record your results in an appropriate table.
23. Repeat step 3 to step 13 with the unknown concentration of hydrogen peroxide solution, $\mathbf{U}$.
24. Count the number of bubbles produced in 120 seconds. Record the result in (a)(iii).
(iii) State the number of bubbles produced in 120 seconds for $\mathbf{U}$.
(iv) Using your results from (a)(ii) and (a)(iii) estimate the concentration of $\mathbf{U}$.
(v) The volume of oxygen was measured by counting the number of bubbles.

Suggest why this is a source of error.
$\qquad$
$\qquad$
(vi) Suggest an improvement to the procedure to provide a more accurate measurement of the volume of oxygen produced.
$\qquad$
$\qquad$
$\qquad$
(vii) In the procedure described in step 1 to step 14, the effect of the concentration of hydrogen peroxide on catalase activity was investigated.

Describe how you would modify this procedure to investigate the effect of changing pH on the number of bubbles of oxygen produced.
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$\qquad$
$\qquad$
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$\qquad$
(b) A student carried out another investigation into the effect of temperature on the activity of catalase. Test-tubes containing plant extract and hydrogen peroxide were placed in thermostatically controlled water-baths at five different temperatures for 120 seconds.

The results are shown in Table 1.2.
The activity of the enzyme is shown in arbitrary units (au).
Table 1.2

| temperature $/{ }^{\circ} \mathrm{C}$ | activity of enzyme/au |
| :---: | :---: |
| 15.5 | 3.4 |
| 26.0 | 6.5 |
| 37.5 | 10.1 |
| 45.5 | 8.2 |
| 55.0 | 1.2 |

(i) Plot a graph of the data in Table 1.2 on the grid in Fig. 1.5.

Use a sharp pencil for drawing graphs.


Fig. 1.5
(ii) Use your graph to find the activity of the enzyme when the temperature was $30.1^{\circ} \mathrm{C}$.
activity of the enzyme = .................................................... au [1]
(iii) Suggest an explanation for the difference in activity of the enzyme between $37.5^{\circ} \mathrm{C}$ and $55.0^{\circ} \mathrm{C}$.
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$\qquad$
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2 M1 is a slide of a stained transverse section through a plant stem.
(a) Set up the microscope so that you can observe the section on M1.

Use a sharp pencil for drawing.
(i) Draw a large plan diagram of the whole section on M1.

Your drawing should show the correct shapes and proportions of the different tissues.
Use one ruled label line and label to identify the epidermis.
(ii) Observe the cells in the central tissue of the section on M1.

Select four adjacent, touching cells of the central tissue.
Each cell must touch at least two of the other cells.

- Make a large drawing of this group of four cells.
- Use one ruled label line and label to identify the cell wall of one cell.
(b) Fig. 2.1 is a photomicrograph of a stained transverse section through a stem of a different type of plant.


Fig. 2.1

Identify the observable differences between the section on M1 and the section shown in Fig. 2.1.

Record the observable differences in Table 2.1.
Table 2.1

| feature | M1 | Fig. 2.1 |
| :---: | :---: | :---: |
|  |  |  |
|  |  |  |

(c) Fig. 2.2 is a photomicrograph of the same plant stem section that is in Fig. 2.1.


Fig. 2.2
You will need to use the grid to find the area of the vascular bundle labelled $\mathbf{A}$ and the area of the vascular bundle labelled B in Fig. 2.2.

Each square of the grid is $1 \mathrm{~cm}^{2}$.
Some of the squares are not completely filled with vascular bundle tissue.
(i) Describe the method you will use to decide which of these squares to include.
$\qquad$
$\qquad$
$\qquad$
(ii) Use the grid to estimate the area of the vascular bundle labelled $\mathbf{A}$ and the area of the vascular bundle labelled B in Fig. 2.2.
area of vascular bundle $\mathbf{A}=$ $\mathrm{cm}^{2}$
area of vascular bundle $\mathbf{B}=$ $\mathrm{cm}^{2}$
(iii) Calculate the area of vascular bundle $\mathbf{A}$ as a percentage of the area of vascular bundle $\mathbf{B}$. Show your working.

## answer $=$

\% [2]
[Total: 18]

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